

Synthesis of Accurate Distance Models of the Primary Donor - Primary Acceptor Pair of Bacterial Photosynthetic Reaction Centres

Maxwell J. Crossley,* Andrew C. Try and Robin Walton

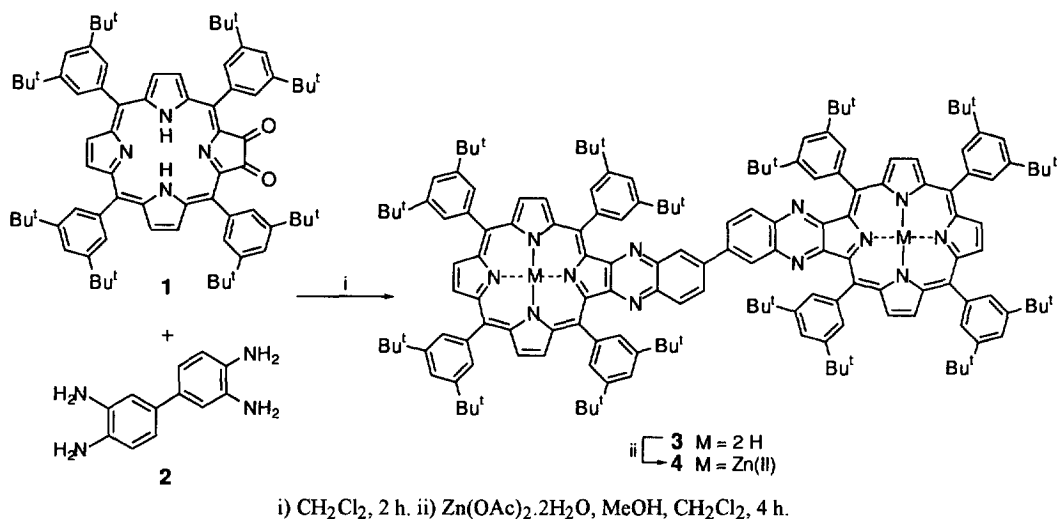
School of Chemistry, The University of Sydney, NSW 2006, Australia

Abstract: The synthesis of **4** and **9**, in which two porphyrins are joined by a biquinoxaliny bridge and by a quinoxaline Tröger's base, respectively, provide systems which have interporphyrin centre-centre and edge-edge distances that very closely match those between the primary donor - primary acceptor tetrapyrrolic pair in bacterial photosynthetic reaction centres. Copyright © 1996 Elsevier Science Ltd

A number of recent studies involving bis-porphyrin systems have been directed towards modelling of features revealed in the X-ray crystal structures of the photosynthetic reaction centre (PRC) of the purple bacterium *Rhodospseudomonas viridis* and of *Rhodobacter sphaeroides*.¹⁻⁶ There is a very similar arrangement of the macrocyclic components that brings about charge separation in the two systems. In particular there is a special pair of bacteriochlorophyll of approximate C_2 symmetry which is the primary electron donor and a nearby bacteriopheophytin which is the primary acceptor.⁷ A number of factors relating to this donor-acceptor pair will influence the charge separation process; some of these are the redox driving forces, modulating properties of nearby groups, the nature of the intervening medium and the relative geometry of the pair. The X-ray structure revealed that the chromophores of this donor-acceptor (D-A) pair on the redox-active L-side have a centre-centre distance of 16.5 Å and an intermolecular edge-edge shortest distance of 9.5 Å.⁸

Previously no synthetic systems have been described that closely model these intermacrocylic distances. We now report the synthesis of the bis-metalloporphyrin systems, **4** and **9**, with centre-centre and macrocycle edge-edge distances that match very closely those found between the primary donor - primary acceptor tetrapyrrolic pair in PRC's.

We have shown earlier that the porphyrin-2,3-dione **1** reacts with 1,2,4,5-benzenetetramine to give extended porphyrin systems⁹ and with *o*-phenylenediamine to give quinoxalinoporphyryns.¹⁰ As expected, other diamino-arenes react with the porphyrin-2,3-dione **1** to give corresponding quinoxalinoporphyryns. We used this reaction to synthesise the biquinoxaliny-bridged bis-porphyrin **3**¹¹ in 89% yield by condensing 2 equivalents of the porphyrin-2,3-dione **1** with 3,3'-diaminobenzidine **2** (Scheme 1). Treatment of **3** with excess zinc(II) acetate dihydrate afforded the dizinc(II) bis-porphyrin **4** in 86% yield. Dicopper(II) and dinickel(II) derivatives of **3** have also been prepared. Compounds **3** and **4** are shown in Scheme 1 in their *anti*-conformation ($\chi = 180^\circ$) about the biaryl linking bond.



Scheme 1

The two porphyrins can rotate with respect to each other about the biaryl bond of **4** but this has only a very small effect on the edge-edge or centre-centre distances in the compound. Molecular modelling of bis-porphyrin **4** using the energy-minimization program SPARTAN¹² yielded an unconstrained minimized structure with the interplane angle $\chi = 32^\circ$ and a centre-centre distance of 18.8 Å and an edge-edge distance of 10.4 Å. Molecular modelling of bis-porphyrin **4** in its three extreme conformations, the *syn*-planar ($\chi = 0^\circ$), the *anti*-planar ($\chi = 180^\circ$), and the orthogonal ($\chi = 90^\circ$), yielded the interporphyrin centre-centre and edge-edge distance ranges given in Table 1.

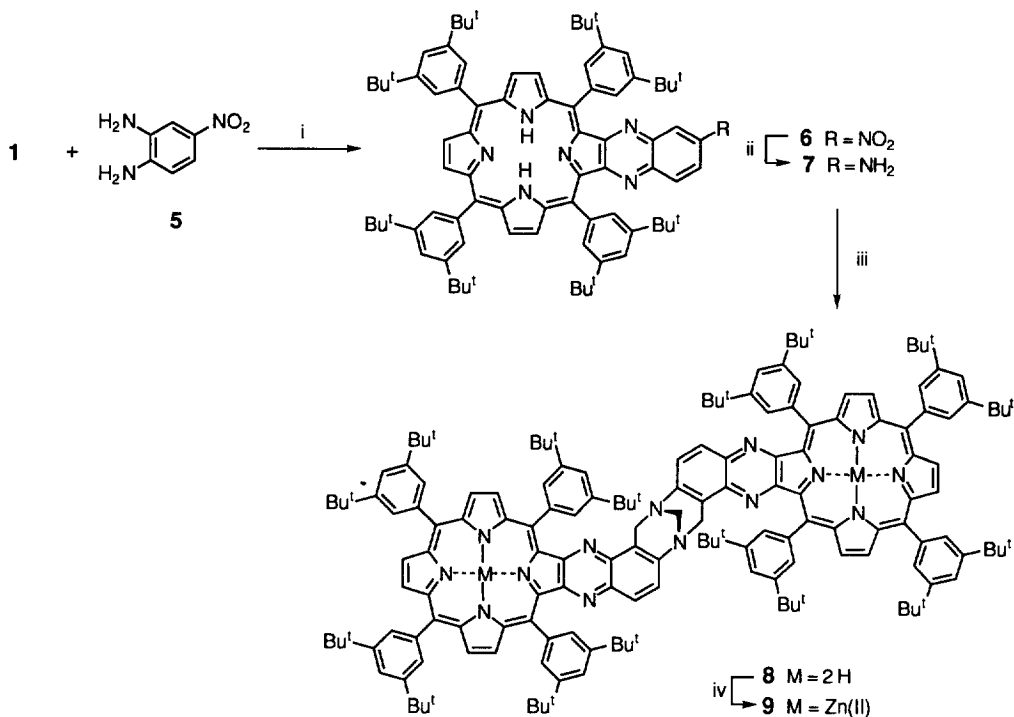
Table 1. Distance (Å) between macrocyclic rings in a PRC and in compounds **4** and **9**

System	Edge-Edge ^a	Centre-Centre
D-A Pair in PRC ⁸	9.5	16.5
Biquinoxaliny Bis-porphyrin 4 ($\chi = 0 - 180^\circ$)	10.4-11.2	18.4-19.5
Tröger's Base Quinoxalino-porphyrin 9	9.8	16.8

^a The shortest distance between β -pyrrolic positions of the two (reduced) porphyrins.

In order to fully constrain the geometry it is necessary to double-pin each of the porphyrins to a rigid linker. The diazocine bridge of Tröger's base achieves this fixation. We have shown previously that a porphyrin-appended Tröger's base analogue results from condensation of a 2-aminoporphyrin with formaldehyde; this bis-porphyrin has a centre-centre distance of only 8.3-8.9 Å.¹³ In order to more closely mimic the geometry of the D-A pair in the PRC, we sought to separate the porphyrin rings via an expanded bridge. Condensation of the porphyrin-2,3-dione **1** with 1,2-diamino-4-nitrobenzene **5** gave the nitroquinoxalino-porphyrin **6** (95% yield) which was subsequently reduced with SnCl₂/HCl. The resultant aminoquinoxalino-porphyrin **7** (71% yield) was reacted with formaldehyde to produce the expanded

quinoxaline-bridged Tröger's base analogue **8**¹⁴ in 88% yield (Scheme 2). Metallation with zinc(II) acetate dihydrate produced the dizinc(II) compound **9** in 73% yield. Monozinc(II), dinickel(II), dipalladium(II) and mixed metal zinc(II)-gold(III) derivatives of **8** have also been prepared.



i) CH₂Cl₂, pyridine. ii) SnCl₂, HCl, CH₂Cl₂. iii) CH₂O, HCl, EtOH, THF, reflux, 18 h.

iv) Zn(OAc)₂·2H₂O, MeOH, CH₂Cl₂, reflux.

Scheme 2

Molecular modelling¹² generated an energy-minimized structure for **9** in which the two quinoxalinoporphyrin units are arranged at an 96° interplanar angle with the distances shown in Table 1. This angle lies in the range seen in the X-ray structures of simpler Tröger's base derivatives (92-104°).¹⁵ The centre-centre distance of about 16.5 Å in the dizinc(II) compound **9** has been measured independently by use of bidentate ligands as 'molecular rulers' as will be described elsewhere.¹⁶

Most synthetic models of the D-A pair have been based on the linkage of two metalloporphyrin units but generally they have been linked by a flexible bridge so that the intermacrocycle distance is not controlled.¹⁻⁶ In contrast, the models reported here not only have remarkably similar edge-edge and centre-centre distances to those found in the PRC but also a constrained geometry. Studies of photoinduced electron-transfer in Zn(II)-free base and Zn(II)-Au(III) derivatives of **3** and **8** should provide information about the influence of the intervening medium on the process. The syntheses reported in this paper further illustrate the versatility of the α-dione **1** as a component in building interesting porphyrin arrays.^{9,17}

Acknowledgements

We thank the Australian Research Council for a research grant to M. J. C., and the Australian Government for Post-graduate Research Awards to A. C. T. and R. W.

References and Notes

1. Chardon-Noblat, S.; Sauvage, J.-P.; Mathis, P. *Angew. Chem., Int. Ed. Engl.* **1989**, *28*, 593-595.
2. Osuka, A.; Nakajima, S.; Okada, T.; Taniguchi, S.; Nozaki, K.; Ohno, T.; Yamazaki, I.; Nishimura, Y.; Mataga, N. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 92-95.
3. Osuka, A.; Maruyama, K. *J. Am. Chem. Soc.* **1988**, *110*, 4454-4456.
4. Wasielewski, M.R. *Chem. Rev.* **1992**, *92*, 435-461 and the references cited therein.
5. Meier, H.; Kobuke, Y.; Kugimiya, S. *J. Chem. Soc., Chem. Commun.* **1989**, 923-924.
6. Sessler, J.L.; Johnson, M.R. *Angew. Chem., Int. Ed. Engl.* **1987**, *26*, 678-680.
7. Deisenhofer, J.; Epp, O.; Miki, K.; Huber, R.; Michel, H. *J. Mol. Biol.* **1984**, *180*, 385-398.
8. Boxer, S.G.; Goldstein, R.A.; Lockhart, D.J.; Middendorf, T.R.; Takiff, L. *J. Phys. Chem.* **1989**, *93*, 8280-8294.
9. Crossley, M.J.; Burn, P.L. *J. Chem. Soc., Chem. Commun.* **1987**, 39-40; Crossley, M.J.; Burn, P.L. *J. Chem. Soc., Chem. Commun.* **1991**, 1569-1571; Crossley, M.J.; Govenlock, L.J.; Prashar, J.K. *J. Chem. Soc., Chem. Commun.* **1995**, 2379-2380.
10. Crossley, M.J.; Burn, P.L.; Langford, S.J.; Pyke, S.M.; Stark, A.G. *J. Chem. Soc., Chem. Commun.* **1991**, 1567-1568.
11. All new compounds have been adequately characterized. Selected data for **3**: Anal. Calcd for C₁₆₄H₁₉₀N₁₂: C, 84.6; H, 8.2; N, 7.2. Found: C, 84.6; H, 8.5; N, 7.4; *m/z* 2329.
12. SPARTAN version 4.0, Wavefunction, Inc, Irvine, CA, USA, 1995.
13. Crossley, M.J.; Hambley, T.W.; Mackay, L.G.; Try, A.C.; Walton, R. *J. Chem. Soc., Chem. Commun.* **1995**, 1077-1079.
14. Selected data for **8**: Anal. Calcd for C₁₆₇H₁₉₄N₁₄•0.5CHCl₃: C, 81.9; H, 8.2; N, 8.0. Found: C, 81.7; H, 8.2; N, 7.8; *m/z* 2524.
15. Cowart, M.D.; Sucholeiki, I.; Bukownik, R.R.; Wilcox, C.S. *J. Am. Chem. Soc.* **1988**, *110*, 6204-6210.
16. Crossley, M.J.; Mackay, L.G., unpublished results.
17. Crossley, M.J.; Burn, P.L.; Langford, S.J.; Prashar, J.K. *J. Chem. Soc., Chem. Commun.* **1995**, 1921-1923.

(Received in UK 21 June 1996; accepted 26 July 1996)